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Key indicators

Single-crystal X-ray study T = 296 KMean $\sigma(\text{C}-\text{C}) = 0.003 \text{ Å}$ R factor = 0.046 wR factor = 0.123 Data-to-parameter ratio = 13.6

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4-Chloro-2-[(4-phenylthiazol-2-yl)hydrazonomethyl]phenol: a polymorph with Z' = 2

The title compound, β -C₁₆H₁₂ClN₃OS, crystallizes with two independent molecules in the asymmetric unit, both of which show a similar twisted conformation. It is polymorphic with α -C₁₆H₁₂ClN₃OS [Çaliskan, Daĝ, Yüksektepe, Cukurovali & Murat (2006). *Acta Cryst.* E**62**, o4889–o4890]; both forms crystallize in the same space group. The two independent molecules in β -C₁₆H₁₂ClN₃OS are linked into dimers by pairs of N-H···N hydrogen bonds.

Comment

We have recently reported the crystal structure of 4-chloro-2-[(4-phenyl-thiazol-2-yl)-hydrazonomethyl]phenol, $C_{16}H_{12}Cl-N_3OS$ (Çalışkan *et al.*, 2006), which is hereafter described as the α modification of this material. We now report a second crystalline polymorph, (I), hereafter called β -C₁₆H₁₂ClN₃OS. Both polymorphs crystallize in space group $P2_1/c$; in the α form there is one independent molecule in the asymmetric unit, and in the β form there are two (Fig. 1).



The two molecules in (I) adopt similar conformations. The main backbones of both molecules are close to planar and in both cases the phenyl ring is twisted in the same sense (Table 1). The dihedral angle between the N1 and C1 rings is 29.13 (14)°; the equivalent value for the N4 and C17 rings is 30.25 (13)°. In α -C₁₆H₁₂ClN₃OS (Çalışkan *et al.*, 2006) the unique molecule is almost planar.

Both molecules in (I) adopt an *E* geometry about the azomethine C—N double bond, with N2–N3–C10–C11 = 177.8 (2)° and N5–N6–C26–C27 = -175.5 (2)°. The *E* configuration of the double bond allows the formation of an intramolecular O–H···N hydrogen bond in both molecules (Table 2). Adjacent molecules are then linked into dimeric pairs by the two N–H···N bonds, resulting in an R_2^2 (8) loop (Bernstein *et al.*, 1995; Yüksektepe *et al.*, 2005). The dimers are arranged into (001) layers (Fig. 2).

Experimental

Thiosemicarbazide (0.9113 g, 10 mmol) was added in portions to a stirred solution of 5-chlorosalicylaldehyde (1.5657 g, 10 mmol) in 30 ml of ethanol. Subsequently, a solution of phenacyl chloride (1.5459 g, 10 mmol) in 20 ml of ethanol was added dropwise. After the addition of the haloketone, the temperature was kept at 323–

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Figure 1

The asymmetric unit of (I) showing 40% displacement ellipsoids (arbitrary spheres for the H atoms). The hydrogen bonds are indicated by dashed lines.





328 K for 2 h more. After cooling to room temperature the solution was made alkaline with an aqueous solution of NH_3 (5%), and a light yellow precipitate separated. The precipitate was filtered off, washed with aqueous NH₃ solution several times and dried in air. Diffraction quality crystals of (I) were obtained by slow evaporation of its ethanol solution. Yield 85%, m.p. 505 K. IR (cm⁻¹): 3280 ν (O-H), 3120 v (N-H), 1614 v(C=N azomethine), 1604 v(C=N thiazole), 676 ν (C-S-C thiazole). ¹H NMR (CDCl₃, p.p.m.): δ 6.96 (s, 1H, thiazole), 7.20-7.89 (m, 8H, aromatics), 8.31 (s, 1H, azomethine), 10.40 (s, 1H, -NH-), 12.20 (s, 1H, -OH). ¹³C NMR shifts (CDCl₃, p.p.m.): δ 125.05 (C1), 156.41 (C2), 119.74 (C3), 131.57 (C4), 123.93 (C₅), 130.39 (C₆),139.38 (C₇), 169.69 (C₈), 105.31 (C₉), 152.2 (C₁₀), 136.22 (C₁₁), 126.69 (C₁₂), 127.33 (C₁₃), 129.37 (C₁₄).

Crystal data

C ₁₆ H ₁₂ ClN ₃ OS
$M_r = 329.80$
Monoclinic, $P2_1/c$
$a = 10.8757 (4) \text{\AA}$
b = 8.2302(5) Å
c = 34.9645 (13) Å
$\beta = 101.144 \ (3)^{\circ}$
$V = 3070.6 (2) \text{ Å}^3$

Data collection

Stoe IPDS-2 diffractometer ω scans Absorption correction: integration (X-RED32; Stoe & Cie, 2002) $T_{\min} = 0.765, \ T_{\max} = 0.904$

Refinement

Refinement on F^2	$w = 1/[\sigma^2(F_0^2) + (0.0772P)^2]$
$R[F^2 > 2\sigma(F^2)] = 0.046$	where $P = (F_o^2 + 2F_c^2)/3$
$wR(F^2) = 0.123$	$(\Delta/\sigma)_{\rm max} = 0.001$
S = 0.99	$\Delta \rho_{\rm max} = 0.75 \ {\rm e} \ {\rm \AA}^{-3}$
5398 reflections	$\Delta \rho_{\rm min} = -0.70 \text{ e } \text{\AA}^{-3}$
398 parameters	Extinction correction: SHELXL97
H-atom parameters constrained	Extinction coefficient: 0.0204 (13)

Z = 8

 $D_x = 1.427 \text{ Mg m}^{-3}$

 $0.60 \times 0.51 \times 0.31 \text{ mm}$

36136 measured reflections

5398 independent reflections

3603 reflections with $I > 2\sigma(I)$

Mo $K\alpha$ radiation $\mu = 0.39 \text{ mm}^{-1}$ T = 296 KPrism, pale yellow

 $R_{\rm int}=0.064$

 $\theta_{\rm max} = 25.0^{\circ}$

Table 1 Selected torsion angles (°).

C6-C1-C7-N1	29.0 (3)	C18-C17-C23-N4	30.0 (3)

Table 2 Hydrogen-bond geometry (Å, °).

$D - H \cdots A$	D-H	$H \cdot \cdot \cdot A$	$D \cdots A$	$D - \mathbf{H} \cdot \cdot \cdot A$
$N2-H2A\cdots N4$ $N5-H5A\cdots N1$ $O1-H1\cdots N3$ $O2-H2B\cdots N6$	0.86	2.11	2.967 (3)	171
	0.86	2.11	2.962 (3)	170
	0.82	1.94	2.661 (2)	146
	0.82	1.95	2.661 (2)	144

All H atoms were placed in idealized positions (C-H = 0.93 Å, N-H = 0.86 Å, O-H = 0.82 Å) and then treated as riding atoms, with $U_{iso}(H) = 1.2U_{eq}(C,N)$ or $1.5U_{eq}(O)$.

Data collection: X-AREA (Stoe & Cie, 2002); cell refinement: X-AREA: data reduction: X-RED32 (Stoe & Cie, 2002): program(s) used to solve structure: SHELXS97 (Sheldrick, 1997); program(s) used to refine structure: SHELXL97 (Sheldrick, 1997); molecular graphics: ORTEP-3 (Farrugia, 1997); software used to prepare material for publication: WinGX (Farrugia, 1999).

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